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Gas delivery systems. How to ensure uninterrupted high quality gas delivery from the source to the apparatus?

Dr. Gastons Vereskuns Trakai, Lithuania May 7, 2014

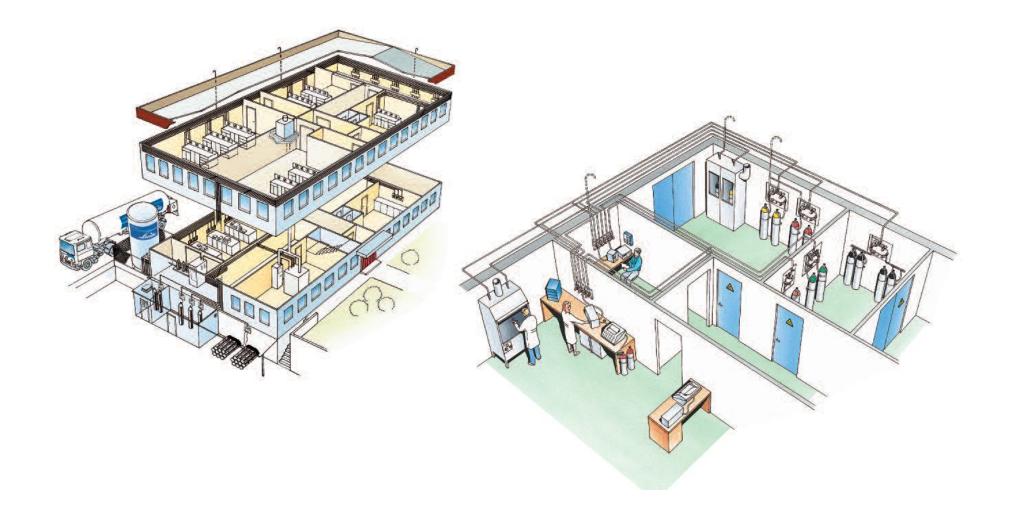


It is a system containing gas pipes, regulators, safety valves and other devices necessary for safe delivery of gases from a source to end user.

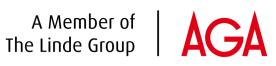
- 1. From a source (gas cylinder, liquid tank, generator)
- 2. Through regulators and pipe system
- 3. To end apparatus (is setting demand for gas and delivery system specification)

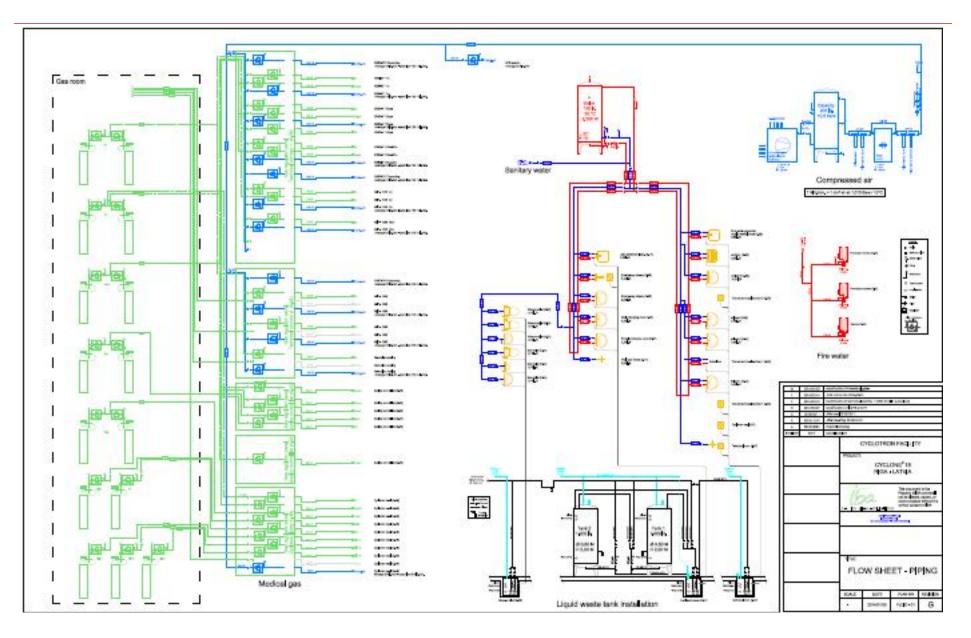


Which gases, how much and where will be used? A Member of The Linde Group



Project of the gas delivery system

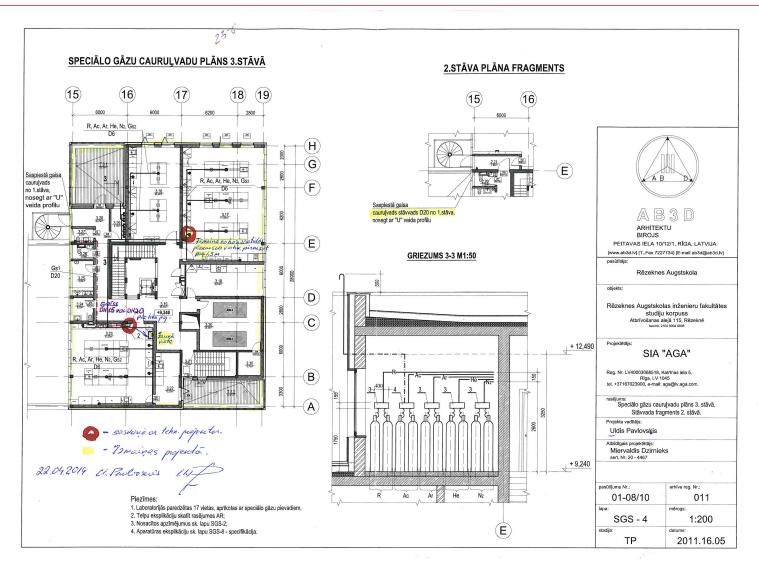




Correct project from the beginning

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Projecting and risk management

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SAFETY ALWAYS AT FIRST!

- Always perform risk analyses.
- Continuous risk management.
- Activities to lower the risks.

The most common risks

- Pressure (>200 bar)
- Gas properties (oxidizing, flammable, toxic, corrosive, etc.)
- Correct materials
- Installation works

Main source of injuries – gas cylinder







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SAFETY ALWAYS AT FIRST!

- Working pressure
- Max working pressure
- Projected pressure
- Test pressure
- Safety valve pressure
- Pressure marks (PN25)

Normal working pressure from regulator/gas source Max pressure from regulator/gas source Max pressure for the gas delivery system Usually safety valve pressure (> max working pressure) (1,43) x Projected pressure Pressure when safety valve opens Designed pressure for materials (regulators, tubes, etc.) PN≥ Projected pressure

Gas delivery systems - Gas purity

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High purity systems – impurities at **ppm** level ppm - parts per million (1 / 1,000,000 or 10-6)

Ultra-High purity systems – impurities at **ppb** level *ppb* - *parts per billion* (1 / 1,000,000,000 or 10-9)



30/04/2014

High purity systems

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<u>Tightness:</u>

 $1x10^{-7}$ cm³/s (gas leak - 1 cm³ He at 1 atm in 4 months)

Main requirements to materials:

- Low or unspecified level of solid particles
- Low leakage of gas (pressure tests)
- Surface Easy to clean / hard to pollute

Materials for pipes:

Stainless steel CQE (Chemistry Quality)

Materials for regulators and valves:

Brass/ stainless steel – Chemically cleaned

Connections:

- Preferably orbital welding
- Swagelok quality compression fittings
- Threads, preferably NPT

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<u>Tightness:</u>

> $1x10^{-9}$ cm³/s (gas leak - 1 cm³ He at 1 atm in 30 years)

Main requirements to materials :

- Very low level of solid particles
- Very low leakage of gas (He leak tests)
- Surface Easy to clean / hard to pollute

Materials for pipes:

Electro polished stainless steel, PQE

Materials for regulators and valves:

Electro polished stainless steel, preferably metal to metal sealings

Connections:

- Orbital welding
- VCR connections
- Welded connections

Pipes – which ones?

Right diameter (pressure drop)

- minimum different pipes (lower costs) 6x1 mm, 12x1 mm, 18x1,5 mm

Suitable for required pressures

Right materials - keep unchanged gas quality from source to end apparatus

- do not produce impurities
- prevent diffusion
- do not react with gas

Special requirements:

- insulated pipes;
- double wall pipes;
- pipes in protective coating, ect.

Stainless steel (SS)

- CQ chemically cleaned, inner surface roughness Ra ~ 1,0 µm
- CQE chemically cleaned, inner surface roughness Ra $\leq 0,4 \,\mu m$
- PQE chemically cleaned, electro polished, inner surface roughness Ra $\leq 0,2 \,\mu$ m,

— Brass— — ?!

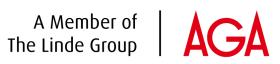
• Chemically cleaned – for medical gases

Polymers (Tellon, PVC II C.)

???!

• Teflon – for aggressive gases

Materials compatibility







Material compatibility.

Gas name	Chemical				12	-								TO .		
	formula	F			-per	bee				8		8		ge		
		DUF	z		db	L S	37	127		ene		2		BSS		
		Aluminum	Buna ^a	8 69 55	Butyl ubber	Carbon steel	Copper	Kel-F*	Moneta	Neoprene ^a	Nylon ^a	Polyethene		Stainless steel	Teflon®	1
		Alt	Bu	8.6	Bu	3	3	Ke	Mo	Ł	Ń	8	PVC	Ste	fel	a second
Acetylene	C ₂ H ₂			# ²				۵		۵					۵	
Aic, synthetic	10000									*			٠			1
Ammonia	NH ₁	۵						۵		۵			٠	۵		-
Argon	Ar															1
Arsine	AsH ₂		۵	۵					۵	۵						_
Boron trichloride	BCI,		-	۵				۵			-	۵	۵.		۵	
Boron trifluoride	BF1	6 ¹			۵			4	B ¹	۵		4	4	6	۵	_
1.3-Butadiene	C4H5									4	-	۵	4			_
n-Butane	C ₄ H ₁₀									4						1
iso-Butane	C ₄ H ₁₀									4						-
1-Butene	C ₄ H ₁									۵.	_					_
cis-2-Butene	C,H,									۵	_					_
iso-Butene	C,H,	-	-							*		4				_
trans-2-Butene	C,H									۵.						-
Carbon dioxide	CO,	-	-	÷						۵	_	4				-
Carbon monoxide	0											4	4			
Carbonyl sulfide	COS		-			*				-				•1		-
Chlorine	d,		۵			4							Δ.			-
Chlorine trifluoride	CIF.								•10					•10		_
Cyanogen	CNCN	-							-	-			-			-
Cyanogen chloride	CNCI		_	_	_				_	_	_	_	_	4		_
Cyclopropane	C ₃ H ₆									۵						_
Deuterium	D ₂			÷							-					-
Diborane	B ₂ H ₆	-	-	÷	-				-	*	۵		-			
Dichlorosilane	SIH,CL,										8				P ¹¹	-
Dimethylamine	(CH ₁),NH									-						
Dimethyl ether																-
2, 2-Dimethylpropane	C2H40 C4H12		•	-	_					_	_	_				_
2, 2-Dimensylpropane	C ₂ H ₂									4						_
Ethene					8						-	4				_
	C ₂ H ₆				۵					٠		۵	٠			_
Ethylene oxide	C2H4O			8	8	4			۵			-	10			
Ethylamine	C ₂ H ₃ NH ₂	4			۵				۵	۵				•		_
Ethyl chloride	C ₂ H ₃ Cl	#1			8	- Ø ¹	# ¹		• ¹	۵						_
Fluorine	F ₂	4		Δ		4	۵	۵						۵	Δ	_
Germane	GeH				-						۵	2				
Helium	He												٠			_
Hydrogen	Hz									٠		٠	112			_
Hydrogen bromide	HBr			4		۵	4						٠	•		_
Hydrogen chloride	HCI			۵		۵		•10					•	•	P ²³	
Hydrogen fluoride	HF			4	-	4		10	۵'		_	. #			P ¹¹	1
Hydrogen selenide	HSe										Δ					1
Hydrogen sulfide	HJS	- # ¹	Δ.	•		4	4	***		*		۵	Δ.	- # ¹	- # ¹¹	_
Krypton	Kr									٠			٠			
Methane	CH									*						_
Methylamine	CH ₃ NH ₂	۵	۵		۵					۵	_		٠			_
Methyl bromide	CH ₁ Br	86				. 0,1								•		_
Methyl chloride	CHICI	0 ⁶							# 1					•		
Methyl fluoride	CH ₁ E									*			٠			-1
Methyl mercaptan	CH ₁ SH				۵	۵				۵			٠			
Neon	Ne												٠			1

Gas name	Chemical				24.20	1000								-		
	formula	Aluminum	Buna [®] N	8855	Butyl rubber	Carbon steel	Copper	Kel-F*	Monel [®]	Neop rene ^a	Nylon*	Polyethene	PVC	Stainless steel	Teflon®	Vi ton*
Nitric axide	NO	•1			۵	# 1	•			۵			٠			
Nitrogen	N ₂															
Nitrogen dioxide	NO ₂	# 1				#1		4				4				
Nitrogen trifluoride	NF ₁	89					# ¹²					1.272		10		
Nitrous axide	N ₂ D		4			4				Δ.						
Octafluoropropane	C,F,		-		2241					100						
Oxygen	0,				۵	۵										
Phosgene	cocl,			۵	٠	4						۵	•	•		
Phosphine	PH,						100000				۵					
Propane	CH.															۵
Propene	C ₂ H ₆								٠	Δ.		۵	٠			
Propyne	C ₂ H ₄									٠				٠		
R 11 Trichlorofluoromethane	CFCI ₂							4								۵
R 12 Dichlorodifluoromethane	CF ₂ Cl ₂		4					4		٠		۵				4
R 13 Chlorotrifiuoromethane	CF ₂ Cl							4			۵		٠			
R 13 b1 Bromotrifluoromethane	CF ₂ Br							۵		+	۵		۵			
R 14 Tetrafluoromethane	CFa							۵			۵					
R 21 Dichlorofluoromethane	CHFCI							۵					٠			
R 22 Chlorodiffuoromethane	CHF ₂ Cl							4				۵	Δ			
R 23 Trifluoromethane	CHF				٠			۵					٠			
R 114 1,2-Dichlorotetrafluorethane	C ₂ Cl ₂ F ₄				۵			4		٠			٠			۵
R 115 Chloropentafluoroethane	C ₂ CIF5							۵		٠						-
R 116 Hexafluoroethane	C ₂ F6							4								
R 142b 1-Chloro-1, 1-difluoroethane	C2H3CIF,							4				۵	A			
R 152a 1, 1-Diffuoroethane	C2H4F2		۵					۵		۵		# ⁷	٠			
Silane	SiH		- 22								۵	200	1002			۵
Silicon tetrafluoride	SIF					Δ3		# ¹⁰	۵1							
Sulphur dioxide	SO ₂			4				# ¹⁰							P ¹¹	
Sulphur hexafluoride	SFg												٠			
Trimethylamine	(CH ₁) ₂ N				۵					۵			٠			
Tungsten hexafluoride	WFg	¹⁷			-	.0	• P ¹²		* ¹⁰					<p*0< p=""></p*0<>		
Vinyl bromide	C ₂ H ₂ Br	۵'			۵	۵					۵	۵	۵			
Vinyl chloride	C ₂ H ₃ Cl	#1			۵	# 1	 #1 									
Vinyl methyl ether	C ₃ H ₆ O									۵	- 107	۵	۵	٠		
Xenon	Xe															

leg	end	1	Avoid in presence of moisture	6	Possible forming of spontaneously	10 < 180 °C (360 °F)
	Good	2	<65% Cu		flammable compounds	11 < 250 °C (480 °F)
4	Fair	3	Avoid in presence of anygen	7	<50 °C (120 °F)	12 < 400 °C (750 °F)
	Avoid	4	Up to 100 bar (1,450 psi)	8	<100 °C (210 °F)	
	Data missing	5	In absence of any gen	9	<150 °C (300 °F)	

Regulators for High purity systems







I 40X HiQ®



W 40B HiQ®

Materials for High purity systems

The Linde Group





V 200 HiQ®



V 50 HiQ®



V 1315 HiQ®



Swagelok compression fittings

Materials and regulators for Ultra-High purity & aggressive / toxic gas systems

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Connections for orbital welding

Welded and VCR connectors

Compression fittings only for purge

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High purity gas delivery system Location

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Ultra-High purity & aggressive / toxic gas systems Location







Cabinets for cylinders storage Ultra-High purity & aggressive / toxic gas systems

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Cabinets for cylinders storage

The Linde Group





Simple cabinet for outdoor cylinders storage.

Openings for ventilation Cylinders' brackets Painting to protect from outdoor environment Lockable doors Available in different sizes: 1 - 2 (50l) **0D600** 600x596x1997 mm

0D1200 1200x596x1997 mm	1 - 4 (50l)
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Cabinets for cylinders storage

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For internal cylinders storage according to DIN EW 14 470-2



- Compulsory ventilation
- Gas detectors
- Electromechanical safety valves
- Cylinders' brackets
- Not for outdoor use



Cabinets for cylinders storage Ultra-High purity & aggressive / toxic gas systems

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Tailor made cabinet for semiconductor gases



- Designed for particular customer
- Technical support
- In-house quality control
- Clean rooms assembly
- Full functional test before dispatch (hardware and software)
- Corresponds to ATEX
- Configurable

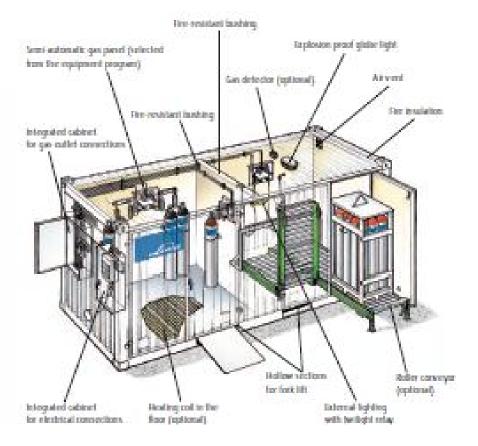
Containers for cylinders storage

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Container 20" (6000x2400x2600)





Technical service and maintenance – Safety and Quality

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Regular technical service:

•Leak tests;

•Pressure tests;

•Exchange of filters;

•Calibration of detectors, etc.

Local regulation:

•annual verification of manometers;•annual tests of liquid tanks, etc.

Emergency service visits:

Safety;

•Economy, etc.

Service and maintenance should be performed by appropriate trained specialists (internal or external). Involvement of technicians with inadequate experience and knowledge can cause serious safety and economical risks.

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THANK YOU!

